

## REACTIONS WITH $\beta$ -CYANOETHYLHYDRAZINE—I

### A ROUTE FOR THE PREPARATION OF PYRAZOLO[1.5-a]PYRIMIDINES AND PYRROLO[1.2-b]PYRRAZOLES

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**Abstract**—Malononitrile (1) reacts with  $\beta$ -cyanoethylhydrazine (2) to yield 5-amino-4-cyano-1- $\beta$ -cyanoethyl-3-cyanomethyl-pyrazole (3). Treatment of 3 with 1% NaOH solution results in the formation of 3-cyano-2-cyanomethyl-4,5,6,7-tetrahydropyrazolo[1.5-a] pyrimidine (6). On the other hand, when 3 was refluxed with acetic acid-hydrochloric acid mixture, the pyrido[3:4:3':4']pyrazolo[1.5-a] pyrimidine (10) was formed.

1,2-Dihydro-5-hydroxy-3-phenylazopyrazolo[1.5-a] pyrimidine (16) was obtained by cyclization of 3,5-diamino-1-( $\beta$ -cyanoethyl)-4-phenylazopyrazole (18), which, in turn, was obtained by the action of 2 on phenylazomalononitrile (17).

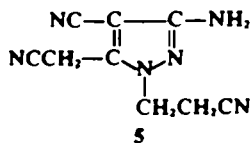
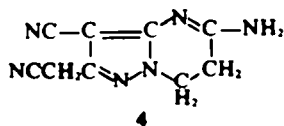
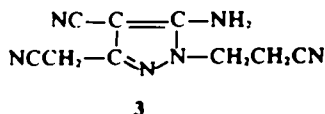
Ethoxymethylenemalononitrile (20) reacts with 2 to yield 3-amino-4-cyano-1-( $\beta$ -cyanoethyl)-pyrazole (21). Compound 21 was readily cyclized into the pyrrolo[1.2-b]pyrazole derivative 25 by the action of either 3% NaOH solution or conc sulphuric acid.

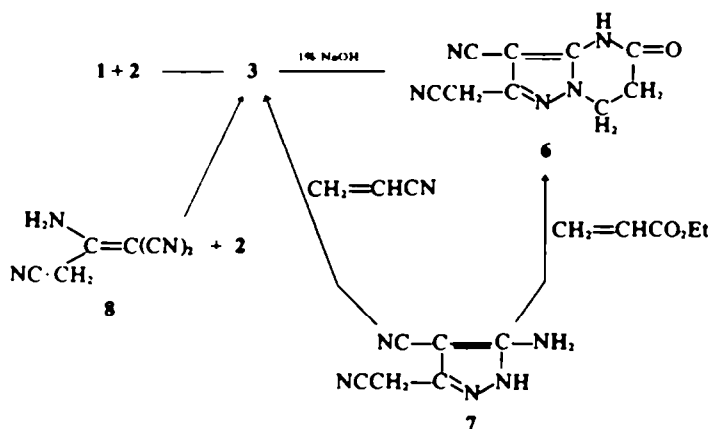
Pyrazolo[1.5-a]pyrimidines have become of considerable interest during the last decade.<sup>1-12</sup> Many of these compounds have proved to be active anticancer,<sup>5-10</sup> antipyretic and anti-inflammatory agents,<sup>8-12</sup> Most of the syntheses reported in the literature involve the reaction of 3-aminopyrazoles with a 1,3-bifunctional reagent. In continuation of previous work on the synthesis of pyrazolo[1.5-a]pyrimidines<sup>3,4</sup> a new route for the synthesis of this class of compounds, via the reaction of malononitrile and its derivatives with  $\beta$ -cyanoethylhydrazine, is reported in this paper. Thus, treatment of malononitrile (1) with  $\beta$ -cyanoethylhydrazine (2), in ethanol solution, resulted in the formation of a product of molecular formula  $C_8H_8N_6$ .

The IR spectrum of the product showed absorption corresponding to amino, unconjugated and conjugated cyano groups. Three structures are possible for this product (cf 3-5). Structure 5 was eliminated since the reaction product was readily converted into 3-cyano-2-cyanomethyl-4,5,6,7-tetrahydropyrazolo[1.5-a]pyrimidine-5-one (6) on treatment with 1% NaOH solution. Compound 6 was also obtained by the action of ethyl acrylate on 5-amino-4-cyano-3-cyanomethylpyrazole (7). Moreover, 3 was also obtained by the action of acrylonitrile on 7.

Structure 3 was preferred over the possible isomeric 4 since its UV spectrum is similar to that reported for 7 and its 1-phenyl derivatives<sup>13,14</sup> and different from that of the pyrazolopyrimidine derivative (6).

The formation of 3 from 1 and 2 may proceed via





malononitrile dimer [1,1,3 - tricyano-2- aminopropene (8)]. This finds support from the fact that 3 is also formed, in a better yield, by the interaction of 2 and 8. This is comparable to the reaction of 1 with hydrazines.<sup>13,14</sup>

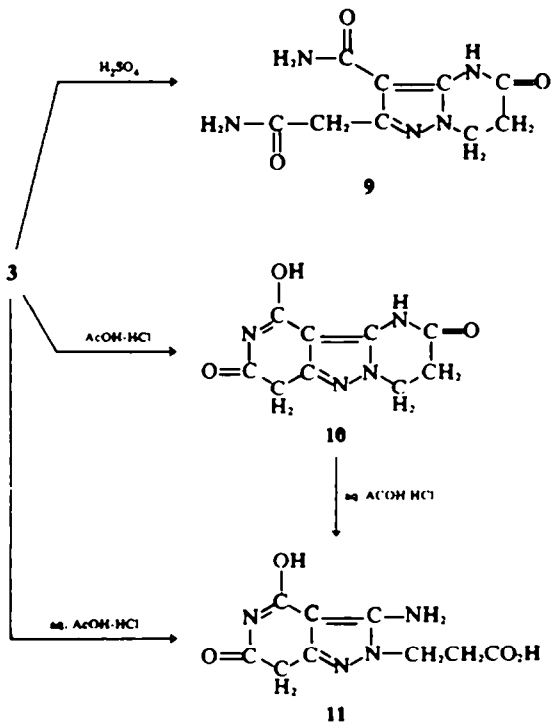
Compounds 3 could be cyclized under different conditions to form different polycyclic compounds. Thus, when 3 was treated with conc sulphuric acid it was converted to the diamide 9. On the other hand, the pyrido [3:4:3':4']pyrazolo[1.5 - a]pyrimidine derivative (10) was formed upon treatment of 3 with acetic acid-hydrochloric acid mixture. However, when 3 was refluxed for a long period with aqueous acetic acid and hydrochloric acid, 3-amino-2 $\beta$  - carboxyethyl-4 - hydroxy-6,7 - dihydropyrazolo[3.4 - c]pyridine-6 - one (11 or possible tautomers) was formed. Compound 11 was also formed by the action of the same reagent on 10. Structures proposed for compounds 9-11 were inferred from their analytical and IR data.

Compound 10 was found to condense readily with benzaldehyde to yield the benzylidene derivative 12, which was also obtained by the cyclization of 3-amino - 7 - benzylidene-2 -  $\beta$  - carboxyethyl-4 - hydroxy-6,7 - dihydropyrazolo[3.4 - c]pyridine-5 - one (13) by the action of acetic acid. Compound 13, in turn, was obtained by the action of benzaldehyde on 11. Compound 10 also coupled with benzenediazonium chloride to yield the corresponding phenylazo derivative 14 (or possible tautomers).

In a previous investigation<sup>4</sup> it has been shown that 3-amino-4 - phenylazo-2 - pyrazolin-5 - one (15) reacts with ethyl acrylate to yield the pyrazolopyrimidine derivative 16. The structure proposed for this compound (16) was mainly based on spectral data. Now, compound 16 has been synthesised as follows: Phenylazomalono-nitrile (17) was treated with 2 at 120° in the absence of a solvent to yield 1- $\beta$  - cyanoethyl-3,5 - diamino-4 - phenylazopyrazole (18). Compound 18 was found to be identical with the product previously obtained by the action of acrylonitrile on 3,5-diamino-4 -

phenylazopyrazole (19). Treatment of 18 with acetic acid-hydrochloric acid mixture affects cyclization and hydrolysis of the amino group to yield 16.

The behaviour of ethoxymethylenemalononitrile (20) toward the action of 2 was also investigated as a possible route for the preparation of pyrazolo[1.5 - a]pyrimidines. When 20 was treated with 2 a crystalline product, mp 200°, with molecular formula of C<sub>7</sub>H<sub>7</sub>N<sub>5</sub>, was formed. The IR spectrum of the product showed bands corresponding to amino, unconjugated cyano, and conjugated cyano groups. There are two possible isomeric structures for this product (cf 21 and 22). It has been reported<sup>15</sup> that 21 reacts with hydrazine hydrate to yield, in addition



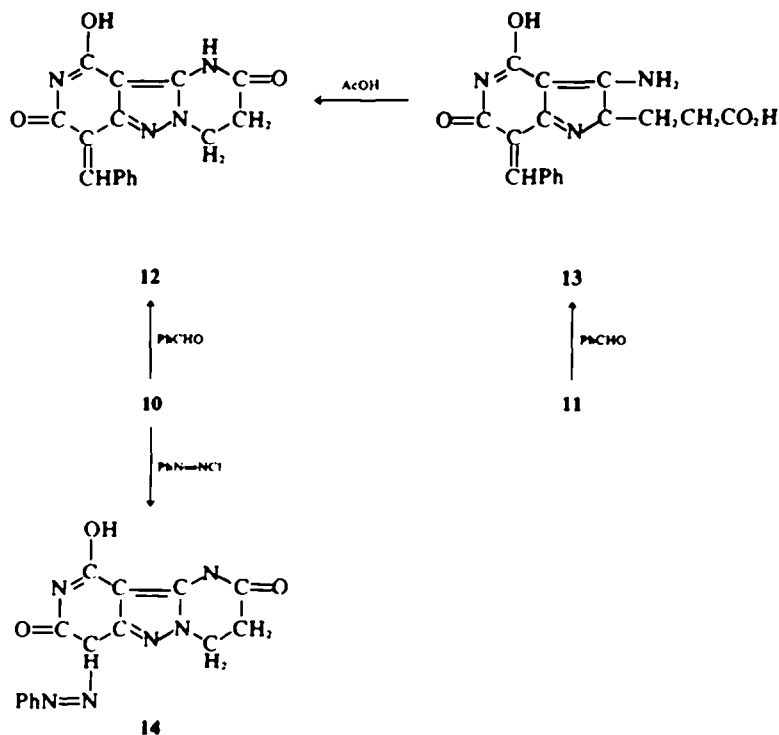
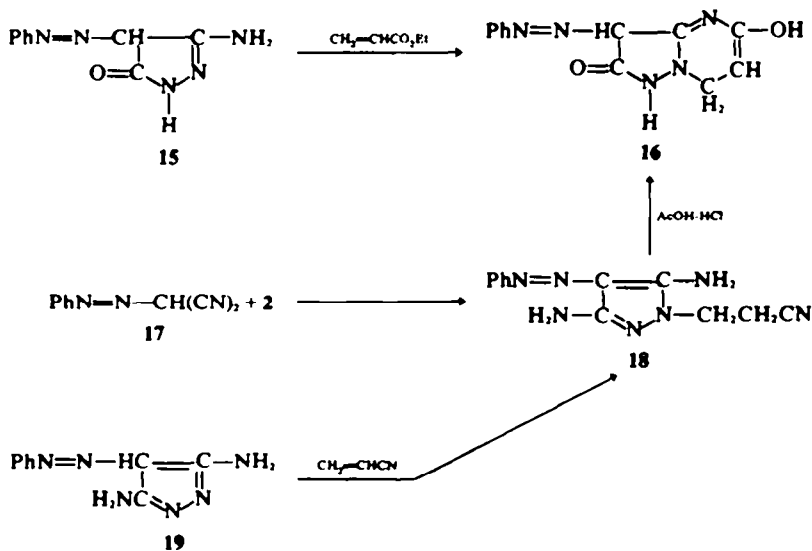
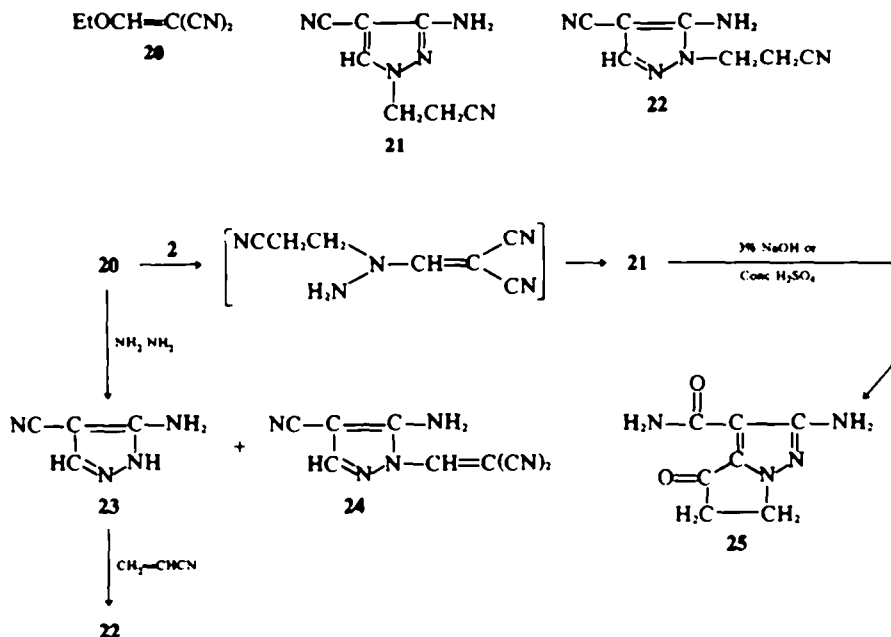


CHART I.

to 3-amino-4-cyanopyrazole (23), 5-amino-4-cyano-1-methylene-malononitrileopyrazole (24). The formation of the latter product may lead to the conclusion that the first step in the reaction of 20 with hydrazines involves condensation of the hydrazine with the ethoxymethylene moiety in 20 followed by addition of the other hydrazine moiety to the cyano group. Taking into consideration the

relative basicities of the N atoms in 2, it is expected that the reaction of 2 with 20 would involve firstly condensation of the more basic NH with the ethoxymethylene group in 20, followed by cyclization via the addition of the NH<sub>2</sub> to the cyano group. This sequence, so illustrated, would lead to compound 21 rather than to 22. However, an independent chemical evidence for this structure proposal





seemed to be mandatory. To establish the structure of the product of the reaction of 2 with 20 (m.p. 200°), compound 23 was treated with acrylonitrile to give a product which can be assigned structure 22, in analogy to the well established structure for the reaction of 7 with acrylonitrile as described above. Since the product of the reaction of 20 with 2 was not identical with 22, it most likely has structure 21.

When 21 was treated with conc sulphuric acid, or with 3% NaOH solution, followed by acidification, it was converted into 25, the structure of which was derived from analytical data, IR spectra and the absence of a singlet around  $\delta$ 2.2 for ring CH in its NMR spectrum. The ease with which 21 was cyclised into 25 can be rationalized in terms of increased reactivity of the ring CH, induced by the presence of the adjacent cyano group, (cf Ref 17).

It is interesting to report here that the cyclization of 21 into 25 constitutes a new way for the preparation of the biologically active<sup>17-19</sup> pyrrolo[1.2-b]pyrazole ring system. Only recently<sup>20</sup> has the synthesis of the alkaloid "Whitasomnine" (a pyrrolo[1.2-b]pyrazole derivative) been reported. Further work on this matter is being carried out to explore the synthetic potentialities of compound 25.

#### EXPERIMENTAL

All m. ps were determined on a micro hot stage and are uncorrected. The IR spectra were recorded with a Hitachi Grating Infrared Spectrometer Model EPI-G3. UV spectra were measured, in water, with a Hitachi 124 spectrophotometer.

**5-Amino-4-cyano-1- $\beta$ -cyanoethyl-3-cyanomethyl-pyrazole (3):**

(1) *From the reaction of 2 with 1.* A mixture of 2 (25 ml) and 1 (33 g) in EtOH (200 ml) was refluxed for 1 h. The mixture was then left to cool gradually to room temp. The solvent was removed *in vacuo* and the remaining oily product was dissolved in hot water. The solid product, obtained on standing, was collected by filtration and crystallized from water.

Compound 3, colourless crystals, m.p. 161°; yield 17.8 g. IR: 3380, 3320 (NH, vibration); 2940, 2920, 2900 (3 CH<sub>2</sub>); 2240 (unconjugated cyano); 2200 (conjugated cyano) and 1660 cm<sup>-1</sup> (NH, deformation). (Found: C, 54.29; H, 4.15; N, 41.77. Calcd for C<sub>8</sub>H<sub>8</sub>N<sub>4</sub>: C, 53.99; H, 4.03; N, 41.98%); UV: no maxima above 220 nm.

(2) *From 7 and acrylonitrile.* A soln of 7 (2.0 g) in pyridine (80 ml) and water (20 ml) was treated with acrylonitrile (1.0 ml) and the mixture was refluxed for 4 h. The solvent was then removed *in vacuo* and the remaining residue was dissolved in hot water. The solid product, obtained on standing, was collected by filtration and proved (m.p. and mixed m.p.) to be 3, yield 2.0 g.

(3) *From 8 and 1.* A soln of 8 (20 g) in EtOH (200 ml) was treated with 2 (25 ml). The mixture was treated as described under 1 and the reaction product (25 g) was identified as 3. Identity was carried out by m.p. and mixed m.p. determination and by IR.

**3-Cyano-2-cyanomethyl-4,5,6,7-tetrahydropyrazolo[1.5-a]pyrimidin-5-one (6)**

(1) *From 3.* Compound 3 (2.0 g) was treated with 1% NaOHaq (100 ml) and the mixture was heated to boiling and the resulting soln was kept at room temp for 3 h. The mixture was neutralized with HCl and then the solvent was removed *in vacuo*. The resulting solid product was dissolved in little hot water and left to cool. The crystals, so formed, were collected by filtration and recrystallized from water.

Compound 6, colourless crystals, m.p. 225°; yield 1.1 g;

IR: 3200, 3050 (NH); 2970, 2920, 2895 (3 CH<sub>2</sub>); 2240 (unconjugated cyano), 2205 (conjugated cyano) and 1680 cm<sup>-1</sup> (CO); UV max  $\lambda$  248 nm. (Found: C, 53.63; H, 3.63; N, 34.64. Calcd for C<sub>8</sub>H<sub>8</sub>N<sub>2</sub>O: C, 53.73; H, 3.51; N, 34.81%).

(2) From 7 and ethyl acrylate. To a soln of 7 (2.0 g) in pyridine (60 ml) and water (10 ml) ethyl acrylate (2.0 ml) was added and the mixture was refluxed for 10 h. The solvent was removed *in vacuo* and the resulting oily product was dissolved in hot water and left to cool. The solid product, so formed, was identified (mp and mixed mp) as 6, yield 1.2 g.

**2-Acetamido-3-carboxamido-4,5,6,7-tetrahydropyrazolo[1.5-a]pyrimidine (9)**

A mixture of 3 (3.0 g) with conc H<sub>2</sub>SO<sub>4</sub> (2.0 ml) was kept at room temp for 1 h. The mixture was then diluted with water (20 ml), neutralized with dil NH<sub>4</sub>OH aq and evaporated *in vacuo* to nearly  $\frac{1}{2}$  of its volume. The solid product, obtained on cooling the mixture, was collected by filtration and crystallized from water.

Compound 9 formed colourless crystals mp 300°; yield 2.2 g; IR: 3397, 3320, 3200 (NH<sub>2</sub> vibrations); 1705, 1670–1660 (CO); 1650–1630 cm<sup>-1</sup> (NH<sub>2</sub> deformations).

**5-Hydroxy-2,3,6,7,8,9-hexahydropyrido[3:4:3':4']pyrazolo[1.5-a]pyrimidin-3,7-dione (10)**

To a mixture of AcOH acid (30 ml) and HCl (8.0 ml), 3.0 g of 3 was added and the mixture was refluxed for 3 h. The solvent was removed *in vacuo* and the residue was treated with a little cold water. The solid product, which separated on standing, was collected by filtration and crystallized from AcOH.

Compound 10 formed yellowish white crystals m.p. 300°; yield 2.7 g; IR: 3590 (OH); 3378, 3170, 3050 (NH group) and 1745–1690 cm<sup>-1</sup> (CO groups). (Found: C, 48.94; H, 3.82; N, 24.87. Calcd for C<sub>8</sub>H<sub>8</sub>N<sub>2</sub>O<sub>3</sub>: C, 49.09; H, 3.66; N, 24.45%).

**3-Amino-2- $\beta$ -carboxyethyl-4-hydroxy-6,7-dihydropyrazolo[3.4-c]pyridin-6-one (11)**

To a mixture of AcOH (40 ml), water (20 ml) and HCl (10 ml; 37.5) 2.0 g of 3 was added, the mixture was refluxed for 16 h. The solvent was then removed *in vacuo* and the remaining solid product was boiled with water (100 ml) and filtered quickly while hot. The insoluble part (1.0 g) was crystallized from AcOH and identified (IR) as 10.

On cooling the filtrate, white crystals were separated and these were collected and recrystallized from water to yield 0.9 g of 11 m.p. 300°; IR: 3580 (OH); 3370, 3190, 3050 (NH<sub>2</sub> vibration); 1720, 1715–1705, 1680 (CO groups) and 1620 cm<sup>-1</sup> (NH<sub>2</sub> deformation). (Found: C, 45.52; H, 4.19; N, 23.71. Calcd for C<sub>8</sub>H<sub>10</sub>N<sub>4</sub>O<sub>3</sub>: C, 45.38; H, 4.23; N, 23.52%).

Compound 11 was also formed on treatment of 10 with aqueous AcOH and HCl mixture under the experimental conditions described above.

**2-Benzylidene-5-hydroxy-2,3,7,8,9-hexahydropyrido[3:4:3':4']pyrazolo[1.5-a]pyrimidin-3,7-dione (12)**

A suspension of 10 in AcOH (40 ml) was treated with benzaldehyde (1.0 ml) and KOAC (4.0 g) and the mixture was refluxed for 3 h. The mixture was then cooled and diluted with water. The solid product, formed on a standing, was collected by filtration and crystallized from

AcOH to yield 1.0 g of 12, yellow crystals m.p. 300. (Found: C, 61.94; H, 3.94; N, 18.00. Calcd for C<sub>11</sub>H<sub>11</sub>N<sub>3</sub>O<sub>3</sub>: C, 62.33; H, 3.92; N, 18.18%; IR: 3190, 3055 (NH<sub>2</sub>); 1715–1690 cm<sup>-1</sup> (CO groups).

**3-Amino-7-benzylidene-2- $\beta$ -carboxyethyl-4-hydroxy-6,7-dihydropyrazolo[3,4-c]pyridine-6-one (13)**

A suspension of 11 (2.0 g) in EtOH (60 ml) was treated with benzaldehyde (1.0 ml) and two drops of piperidine. The mixture was refluxed for 16 h and then left to cool at room temp. The crystals, separated on cooling, were filtered off and recrystallized from dimethylformamide:

Compound 13 formed yellow crystals, m.p. > 300°, yield 1.0 g; IR: 3370, 3225, 3050 (NH<sub>2</sub> vibration); 1720, 1700–1685 (CO groups) and 1640 (NH<sub>2</sub> deformation) cm<sup>-1</sup>. (Found: C, 58.74; H, 4.33; N, 17.36. Calcd for C<sub>11</sub>H<sub>11</sub>N<sub>3</sub>O<sub>3</sub>: C, 58.89; H, 4.32; N, 17.17%).

When compound 13 was refluxed in AcOH for 3 h it was quantitatively converted into 12 (identity was carried out by IR spectra).

**5-Hydroxy-2-phenylazo-2,3,6,7,8,9-hexahydropyrido[3:4:3':4']pyrazolo[1.5-a]pyrimidin-3,7-dione (14)**

To a soln of 10 (3.0 g) in dioxane (80 ml) was added 10% K<sub>2</sub>CO<sub>3</sub> aq (80 ml). The mixture was cooled to 0° and gradually treated with a soln of benzene-diazonium chloride prepared from 1 ml of aniline). The mixture was left in a refrigerator for 10 h and then diluted with water. The solid product, so formed, was collected by filtration and was crystallized from dimethylformamide to yield 2.2 g of 14, golden yellow crystals m.p. > 300°; IR: 35600, 3500 (OH); 3180, 3040 (NH); 1710–1665 (CO) and 1610 cm<sup>-1</sup> (C=N). (Found: C, 55.46; H, 4.08; N, 26.17. Calcd. for C<sub>11</sub>H<sub>11</sub>N<sub>3</sub>O<sub>3</sub>: C, 55.55; H, 3.73; N, 25.92%).

**3,5-Diamino-1-( $\beta$ -cyanoethyl)-4-phenylazopyrazole (18)**

A mixture of 18 (2.0 g) and ( $\beta$ -cyanoethylhydrazine (1.0 ml) was heated at 120° (bath temp) for 3 h. The mixture was then allowed to cool and dissolved in EtOH. The crystals separated out on cooling were recrystallized from EtOH and identified (m.p. and mixed mp) as XVIII, yield, 2.0 g.

**1,2-Dihydro-5-hydroxy-3-phenylazopyrazolo[1.5-a]pyrimidin-2-one (16)**

To a suspension of 18 (2.0 g) in AcOH (30 ml), 27.5% HCl (2 ml) was added and the mixture was refluxed for 3 h. The solid product was collected by filtration while the soln was still hot and crystallized from dimethylformamide and proved to be 16° by IR, yield, 1.8 g.

**3-Amino-4-cyano-1- $\beta$ -cyanoethylpyrazole (21)**

To a sol of 20 (10 g) in EtOH (100 ml), 10 ml of 2 was added. The mixture was heated under reflux for 2 h. The solvent was then removed *in vacuo* and the remaining solid product was crystallized from water.

Compound 21 formed colourless crystals, m.p. 200° (dec) yield 1475 g; IR: 3395, 3338, 3180 (NH, vibration); 2970, 2920 (CH<sub>2</sub>), 2230 (unconjugated cyano); 2205 (conjugated cyano) and 1640 cm<sup>-1</sup> (NH, deformation). (Found: C, 51.91; H, 4.39; N, 43.27. Calcd for C<sub>7</sub>H<sub>7</sub>N<sub>3</sub>: C, 52.16; H, 4.39; N, 43.46%).

**5-Amino-4-cyano-1- $\beta$ -cyanoethylpyrazole (22)**

To a sol of 20 (13 g) in pyridine (100 ml) and water (50 ml), acrylonitrile (6 ml) was added. The mixture was

then treated with one drop of conc KOH<sub>aq</sub> and the remaining oily product was dissolved in hot EtOH. On cooling, white crystals were separated and these were collected and recrystallized from EtOH to yield 13 g of **22**, mp 131°; IR: 3390, 3340, 3180 (NH, vibration); 2980, 2925 (2 CH<sub>2</sub>); 2235 (unconjugated cyano); 2210 (conjugated cyano) and 1630 cm<sup>-1</sup> (NH, deformation). (Found: C, 51.98; H, 4.39; N, 43.29. Calcd for C<sub>8</sub>H<sub>7</sub>N<sub>3</sub>: C, 52.16; H, 4.39; N, 43.46%).

2-Amino-3 - carboxamido-5,6 - dihydropyrrolo[1.2 - b]pyrazol-4 - one (**25**)

(1) *By the action of dilute sodium hydroxide solution on 21.* A mixture of **21** (3 g) and 3% NaOH<sub>aq</sub> (100 ml) was refluxed until no more ammonia odour could be detected (3 h). The mixture was then left to cool; neutralized with dil HCl and evaporated *in vacuo*. The remaining solid product was rinsed with a little water and collected by filtration to yield 2.4 g of **25** which was recrystallized from water.

Compound **25** formed colourless crystals mp 294°; IR: 3390, 3320, 3170, 3080 (NH, vibrations); 1695, 1670 (CO vibrations); 1620 cm<sup>-1</sup> (NH, deformation). (Found: C, 46.06; H, 5.51; N, 31.28. Calcd for C<sub>8</sub>H<sub>10</sub>N<sub>4</sub>O<sub>2</sub>: C, 46.15; H, 5.52; N, 30.96%).

(2) *By the action of concentrated sulphuric acid on 21.* A mixture of **21** (3 g) and conc 98% H<sub>2</sub>SO<sub>4</sub> (2 ml) was kept at room temp for 2 h. The mixture was then diluted with water (20 ml) and neutralized by the gradual addition of dil NH<sub>4</sub>OH<sub>aq</sub>. The mixture was then evaporated *in vacuo* and the remaining solid product was dissolved in the least amount of hot water and left to cool. The crystals, separated on standing, were collected and identified (m.p. and mixed m.p.) as **25**, yield 2.3 g.

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